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Almond based Electrocatalyst for Fuel Cell Applications

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Background

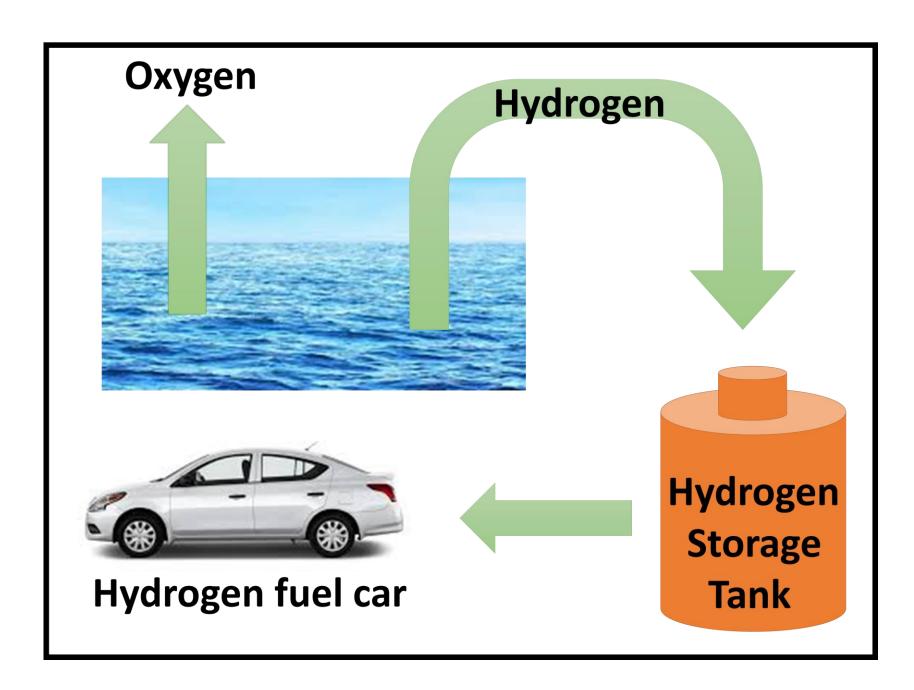
- ☐ The rapid depletion of fossil fuels and growing environmental concerns have created an enormous worldwide demand for alternative clean energy technologies.
- ☐ Hydrogen has a potential to be one of the most effective clean energy sources due to its high energy density of 141.86 MJ/kg and clean combustion.
- ☐ Water electrolysis via photochemical and photoelectrochemical are the promising pathways for sustainable hydrogen production.
- ☐ The water splitting reaction can be expressed chemically as: Total reaction $2H_2O \rightarrow 2H_2 + O_2$ (1.23V)

Challenges

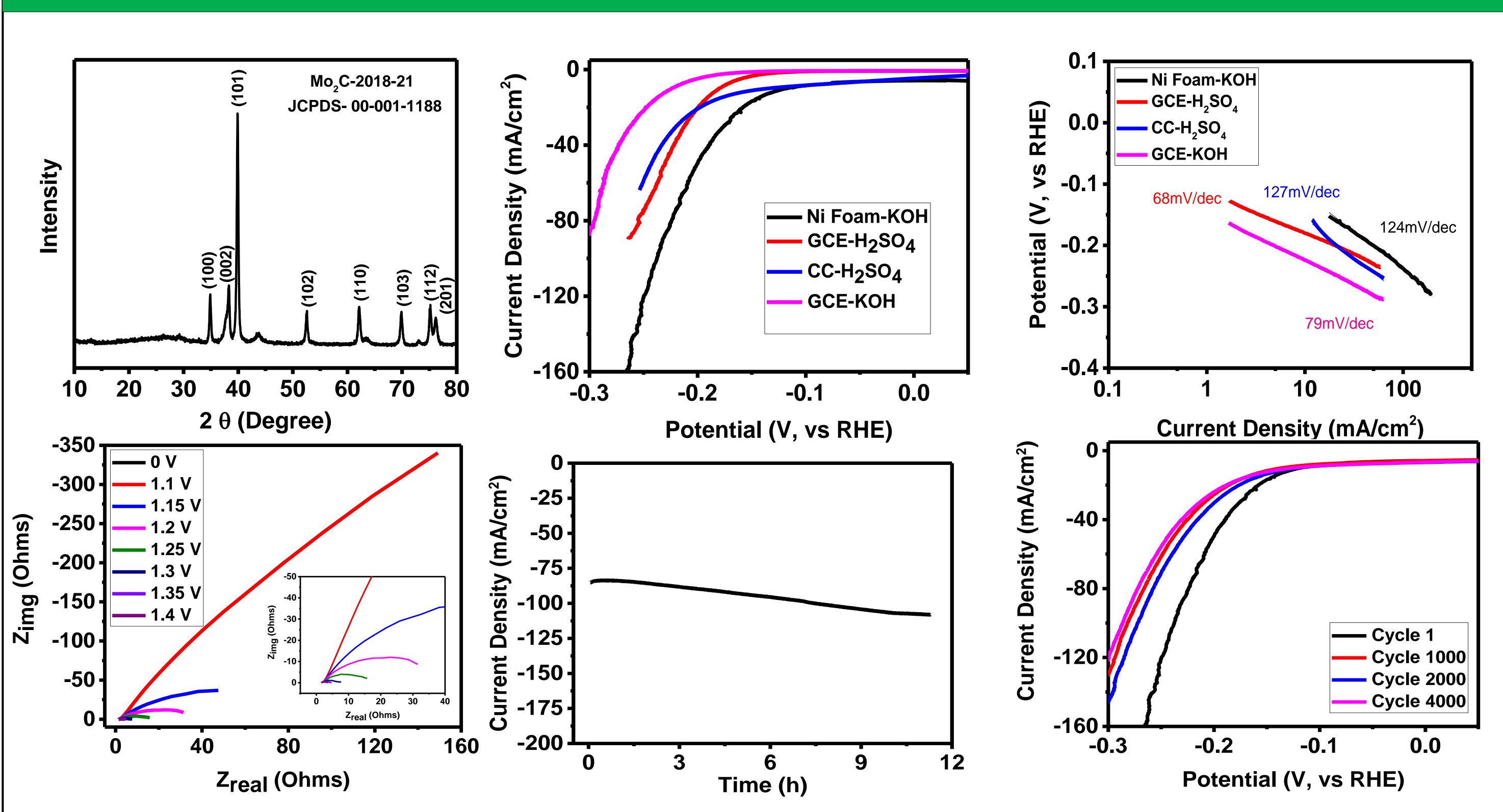
- Regardless of the media in which water splitting takes place, the thermodynamic voltage of water splitting is 1.23 V at 25 °C and 1 atm. However, in fact, we must apply voltages higher than the thermodynamic potential (overpotential) value to achieve electrochemical water splitting.
- An ideal electro-catalyst should provide high current density at a lower overpotential.
- ☐ Platinum is one of the most efficient electro-catalyst for hydrogen evolution, however, its high cost and limited availability has curtailed its use for hydrogen production.

Solution through this research

- ☐ Replacement of Pt with inexpensive, earth-abundant electrocatalysts would be significantly beneficial for clean and efficient hydrogen evolution.
- ☐ We have used earth-abundant materials such as Mo₂C as an alternative to platinum electrocatalyst for hydrogen evolution.



Results and discussion



- \Box We have prepared Mo₂C as a cost effective and biorenewable electrocatalyst capable of performing in both alkaline and acidic media.
- \square Mo₂C on nickel foam produces higher current densities (122 mA/cm² at -0.25 V) compared to the Mo₂C on GCE in acidic media (75 mA/cm² at -0.25 V) and Mo₂C on GCE in alkaline media (22 mA/cm² at -0.25 V).
- ☐ Molybdenum carbide showed an overpotential of 117 mV and 180 mV in alkaline and acidic media, respectively to achieve a current density of 10 mA/cm2
- \Box The long-term stability of the Mo₂C was studied using chronoamperometry and cyclic voltammetry measurements. The performance of Mo₂C is very stable up to about 12 hours of measurement.

Summary

- \triangleright Mo₂C was synthesized using a facile method.
- \succ The excellent electroactivity of Mo₂C is due to its high conductivity, high electroactive surface area, and pure phase structure.
- Mo₂C on nickel foam in basic media, Mo₂C on GCE in basic media and Mo₂C on GCE in acidic media showed Tafel slopes of 124, 79 and 68 mV per decade, respectively.
- \triangleright Mo₂C could be a promising cheap electrocatalyst for hydrogen evolution reaction due to its high electroactivity and durability.

Future work & Acknowledgement

- We are working to develop multifunctional (energy storage and generation) materials using cost effective method.
- > We plan to use other bio-waste materials for the synthesis of electrocatalysts.
- Sincere acknowledgment to the Polymer Chemistry Program, PSU and KINBRE (Grant # P20GM103418) for providing financial and research support.