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Metal-Oxide Frameworks-based Cobalt Oxides as Efficient Electrocatalysts

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 $2 H_2O \rightarrow 2 H_2 + O_2$

Water Splitting

Overview

- **Green energy production via cost-effective ways is one of the main requirements in current days.**
- * Efficient electrocatalysts are needed to produce a green energy, such as hydrogen.
- * Metal-organic frameworks (MOFs)—based materials can become a good candidate for electrocatalysts due to a large surface area and abundant active sites.
- Reaction temperature is an important factor to tune the structure and morphology of the materials, resulting in the different electrocatalytic activities.
- **Co-MOF** materials were synthesized at the various temperatures to improve the efficiency of the water splitting process.

Application Fuel cell vehicles Oxygen tank

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 m^2

Co- MOF 500

----Cycle 1

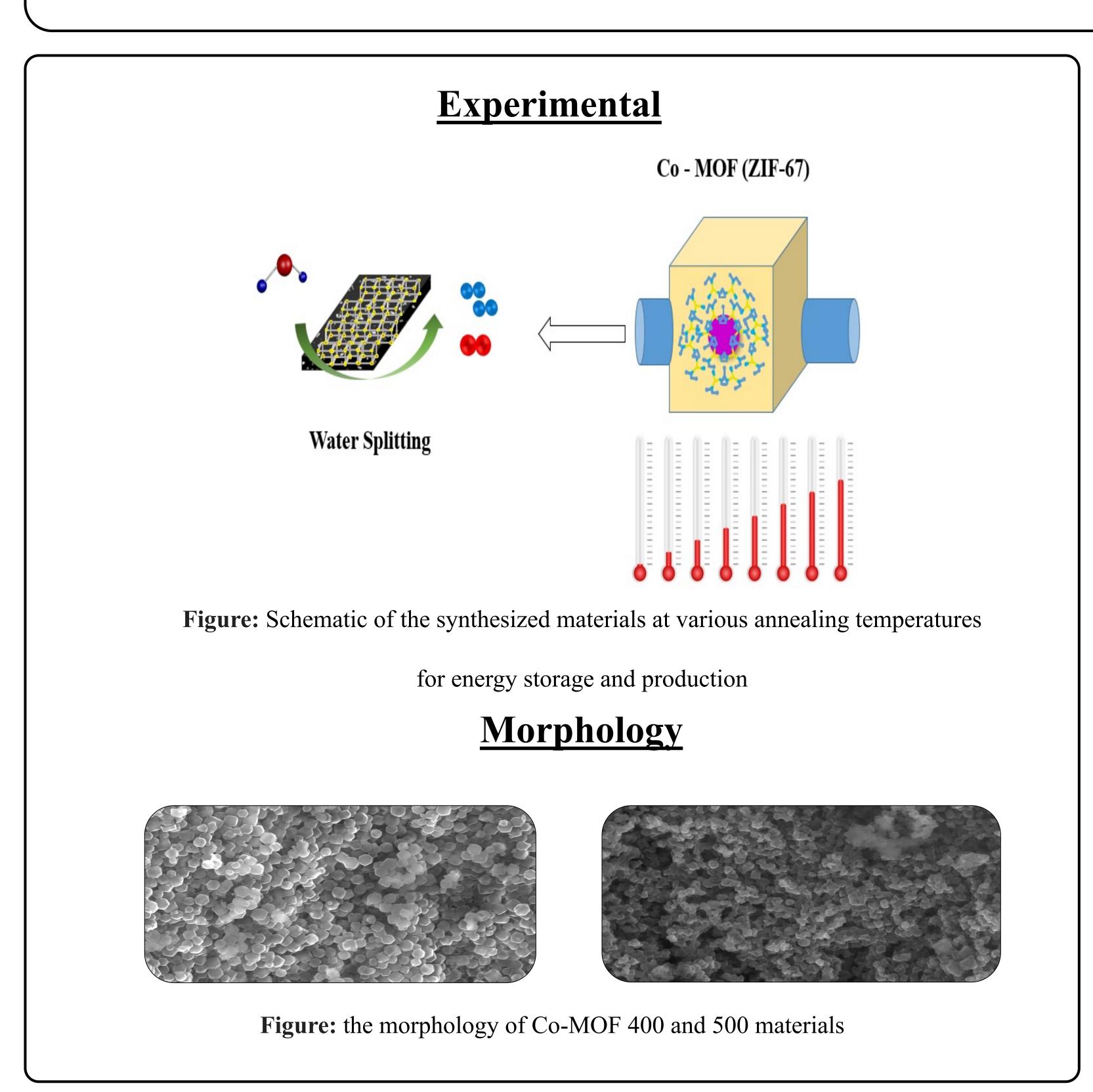
Potential (V, RHE)

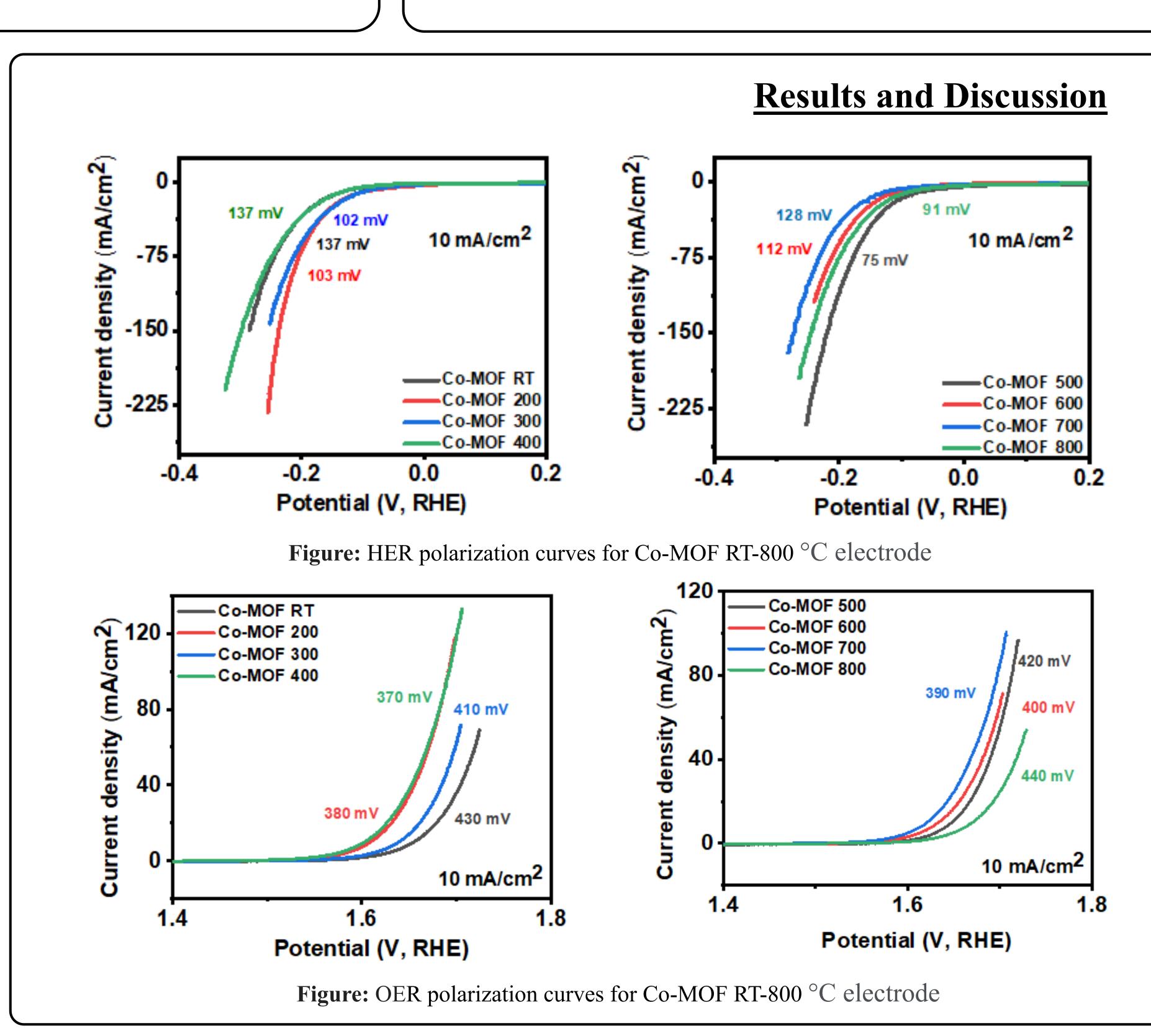
Potential (V, RHE)

Figure: 1 and 1k OER polarization curves for Co-MOF 400

Figure: 1 and 1k HER polarization curves for Co-MOF 500

Cycle 1k





Conclusion

- \clubsuit Metal-oxide frameworks-based cobalt oxides were synthesized at the various temperature (RT -800 °C).
- The Co-MOF electrodes showed low overpotential in the range of 75 to 137 mV to achieve a current density of 10 mA/cm².
- ❖ The Co-MOF electrodes required an overpotential in the range of 370 to 440 mV for oxygen production at 10 mA/cm².
- * These materials showed stable performance for up to 1,000 cycles of cyclic voltammetric studies.

Future Research

❖ Future research focuses on making an electrolyzer to test the performance of electrodes at the twoelectrode system.

Acknowledgement

Co- MOF 400

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